# Effect of additive SrWO<sub>4</sub> on microwave dielectric properties of SrTiO<sub>3</sub>–Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics

Fei Liang • Wenzhong Lu • Dongxiang Zhou • Xiaohong Wang

Published online: 1 September 2007 © Springer Science + Business Media, LLC 2007

Abstract The crystal structure and the properties of a new microwave dielectric ceramics x SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub> Nb<sub>2/3</sub>)O<sub>3</sub> have been investigated. With x=0.025, The new microwave dielectric ceramic achieves the dielectric properties of a dielectric constant  $\varepsilon_r \sim 27.8$ , a  $Q \times f$  value ~26,800, and a  $\tau_f$  value ~7.4 ppm/°C. When the SrWO<sub>4</sub> is added, the sintering temperature of x SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub> Nb<sub>2/3</sub>)O<sub>3</sub> ceramics will fall to 1350 °C, and its  $Q \times f$  value can be improved further and the  $\tau_f$  value becomes smaller. When the SrWO<sub>4</sub> is added by 0.07 mol, the specimen acquires the following microwave properties: a dielectric constant  $\varepsilon_r \sim 30.3$ , a  $Q \times f$  value ~29,500, and a  $\tau_f$  value of approximately –0.4 ppm/°C.

**Keywords** Microwave ceramics · Dielectric properties complex perovskites

## **1** Introduction

The requirement of rapid data transmission has led to an increased interest in the gigahertz region and has prompted research on dielectric materials at microwave frequencies. Communication system relies on dielectric materials with low losses and high-temperature stability. Most recent research of high-Q dielectrics has focused on complex

F. Liang (⊠) · W. Lu · D. Zhou · X. Wang
Department of Electronics Science and Technology,
Huazhong University of Science and Technology,
1037 Luoyu Road,
Wuhan 430074, People's Republic of China
e-mail: afei liang@tom.com

perovskites with permittivity in the range 30–40, and very high  $Q \times f$  values (~100,000). For example, the most widely used perovskites, Ba(Zn<sub>1/3</sub>Ta<sub>2/3</sub>)O<sub>3</sub> (BZT) has  $\varepsilon_r \sim 29$ ,  $Q \times f =$ 165,000 at 11 GHz, and a near zero temperature coefficient of the resonate frequency  $\tau_{\rm f}$  [1]. However, the high cost of tantalum oxide has motivated investigations of lower cost replacements for BZT. The niobium-based analogues of the tantalite perovskites constitute one family of candidate materials. In this research, we examine the structures and microwave properties of Mg based niobate perovskite  $Sr(Mg_{1/3}Nb_{2/3})O_3(SMN)$ .  $Sr(Mg_{1/3}Nb_{2/3})O_3$  microwave ceramics possesses high dielectric constant ( $\varepsilon_r$ =30), highquality factor ( $Q \times f$  value ~35,000 GHz), and negative  $\tau_{\rm f}$  value (-27 ppm/°C) [2]. It is well known that two or more compounds having negative and positive temperature coefficient values employed to form a solid solution or mixed phase are the most promising method to obtain zero temperature coefficient of resonant frequency. Because SrTiO<sub>3</sub> ceramics exhibits dielectric properties of  $\varepsilon_r \sim 270$ ,  $Q \times f$  value ~3,000 GHz and a large positive  $\tau_{\rm f}$  value (1,200 ppm/°C), compensation of temperature coefficient of resonant frequency ( $\tau_{\rm f}$ ) might be obtained by employing solid solution of SrTiO<sub>3</sub> and Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics. In the present study, microwave dielectric ceramics based on the solid solution of SrTiO<sub>3</sub> and Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> were developed, focused in particular on the effect of the SrWO<sub>4</sub> additive on the their microwave properties and sintering temperature. The microwave dielectric properties were investigated. In addition, the X-ray diffraction (XRD) pattern and the scanning electron microscopy (SEM) analysis were also employed to study the microstructure of the ceramics.





## 2 Experimental procedure

#### 2.1 Samples preparation

Specimen powders were prepared by a conventional solidstate reaction technique. High purity (99.9%) SrCO<sub>3</sub>, TiO<sub>2</sub>, MgO, and Nb<sub>2</sub>O<sub>5</sub> were weighed according to the composition *x* SrTiO<sub>3</sub>–(1–*x*)Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub>, where 0.01  $\leq x \leq$ 0.1, and were ground in ethyl alcohol for 4 h in a balling mill with zirconia balls, dried at 80 °C. Prepared powders were calcined at 1200 °C for 4 h in air. After calcinations, the calcined powders were mixed according to the molar fraction (1–*x*)SrTiO<sub>3</sub>–*x* Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> +*y*SrWO<sub>4</sub> and re-milled for 4 h with a 5% solution of PVA as a binder. The dried power were isostatically pressed into pellets 20 mm in diameter and 13–16 mm in height under the pressure of 2×10<sup>7</sup> Pa. After being fired to remove an organic binder, these samples were sintered at temperature 1350–1600 °C for 4 h in air.

## 2.2 Measuring methods

The crystalline phase of the prepared samples was identified by D/MAX-RB X-ray diffraction pattern (XRD). The microstructure observation of the sintered ceramics was performed by means of JSM-5610LV scanning electron microscopy (SEM). The dielectric constants  $\varepsilon_r$  and an unloaded quality values Q in microwave frequency (2– 6 GHz) were measured using the parallel-plate method combined with a work analyzer(Advantest R3767C) and a computer. The temperature coefficient of resonant frequency was obtained by measuring the TE<sub>01 $\delta$ </sub> resonant coefficient at temperatures -20 °C(*f*-<sub>20</sub>) and 60 °C(*f*<sub>60</sub>).

#### **3** Results and discussions

Figure 1 shows the XRD patterns of *x* SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub> Nb<sub>2/3</sub>)O<sub>3</sub> ceramics. It is observed that as the *x* value increased, the XRD patterns of x SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub> Nb<sub>2/3</sub>)O<sub>3</sub> ceramics have not apparent changes and are very similar to that of Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics, which is mainly owing to small SrTiO<sub>3</sub> mol fraction. Most the peaks from *x* SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics could be indexed based on the 1:2 ordered structure, and complex perovskite structures were observed for all compositions [3].

Table 1 Dielectric properties of the x SrTiO\_3 –(1–x)Sr(Mg\_{1/3}Nb\_{2/3}) O\_3 system ceramic.

x value	$d (g/cm^3)$	$\mathcal{E}_{\mathrm{r}}$	$Q \times f$ (GHz)	$ au_{ m f}$ (ppm)
0.01	4.38	26.5	29,500	-23
0.02	4.33	27.2	27,700	-16.3
0.025	4.37	27.8	26,800	7.4
0.03	4.28	29.3	25,800	17.2
0.04	4.29	30.2	23,400	25.8
0.05	4.17	31.5	22,600	36.3
0.06	4.16	32.8	20,200	47.6
0.07	4.19	33.7	19,400	58.9
0.08	4.15	35.2	18,900	70.6

Fig. 2 SEM of 0.025SrTiO<sub>3</sub>-0.975Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub>: (a) without SrWO<sub>4</sub> addition at 1600 °C for 4 h, (b) with SrWO<sub>4</sub> addition at 1350 °C for 4 h



Therefore, the above results indicate that *x* SrTiO<sub>3</sub>–(1-x)Sr (Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics (0.01 $\leq x \leq 0.08$ ) has 1:2 ordered perovskite structure, which is mainly for that most complex perovskite structure ceramics have high quality factor.

The dielectric properties of x SrTiO<sub>3</sub>–(1–x)Sr(Mg<sub>1/3</sub> Nb<sub>2/3</sub>)O<sub>3</sub> ceramics are illustrated in Table 1. As the x value increased from 0.01 to 0.08, the dielectric constants  $\varepsilon_{\rm r}$  increased from 26.5 to 35.2 while the  $Q \times f$  values gradually decreased, it is because that SrTiO<sub>3</sub> ceramics has lower Q value. The  $\tau_{\rm f}$  value varied from –23 to 70.6 as the x value increased from 0.01 to 0.08 for x SrTiO<sub>3</sub>–(1–x)Sr(Mg<sub>1/3</sub> Nb<sub>2/3</sub>)O<sub>3</sub> ceramics. It implied that near zero  $\tau_{\rm f}$  value can be obtained by properly adjusting the x value. After optimization, a new microwave dielectric material 0.025SrTiO<sub>3</sub>–0.975Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> was suggested and exhibited the dielectric constant  $\varepsilon_{\rm r} \sim 27.8$ , the  $Q \times f$  values ~26,800 and the  $\tau_{\rm f}$  value ~7.4 ppm/°C.

It was found that it was difficult to obtain high density *x* SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics even by sintering at temperature as high as 1600 °C. However, the addition of SrWO<sub>4</sub> additives could effectively promote densification by sintering at 1350 °C and high density. The (a) and (b) of Fig. 2 show respectively SEM of 0.025SrTiO<sub>3</sub>–0.975Sr(Mg<sub>1/3</sub>)

Nb<sub>2/3</sub>)O<sub>3</sub> without SrWO<sub>4</sub> addition at 1600 °C for 4 h and with SrWO<sub>4</sub> addition at 1350 °C for 4 h. From the Fig. 2, it was seen that without SrWO<sub>4</sub> addition, even if the 0.025SrTiO<sub>3</sub>–0.975Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> was sintered on 1600 °C for 4 h, the crystal size was small and the sintered ceramic was not dense. While the SrWO<sub>4</sub> was added, the size of ceramics crystal grew larger and the ceramics became more dense.

Figure 3 shows the changes in dielectric constant of 0.025SrTiO<sub>3</sub>-0.975Sr (Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics with the addition of SrWO<sub>4</sub> at 1350 °C for 4 h. The 0.025SrTiO<sub>3</sub>-0.975Sr (Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics has a dielectric constant of about 27.8, and when SrWO<sub>4</sub> was added, the dielectric constant increased initially because of increasing density. Over 0.07 mol SrWO<sub>4</sub>, the dielectric constant decreased steadily with increasing SrWO<sub>4</sub> because of increasing SrWO<sub>4</sub> phase whose dielectric constant ( $\varepsilon \sim 9$ ) is lower than that of Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> and SrTiO<sub>3</sub>.

Figure 4 shows changes in  $Q \times f$  values with different SrWO<sub>4</sub> mol fraction at 1350 °C for 4 h. From the Fig. 4, the  $Q \times f$  values slowly increased with the SrWO<sub>4</sub> addition firstly up to the 0.03 SrWO<sub>4</sub> mol fraction. It is mainly because that the density was increased due to better sinterability with SrWO<sub>4</sub>. When the addition of SrWO<sub>4</sub> is over 0.03 mol, the  $Q \times f$  values rapidly increase, which may



Fig. 3 Dielectric constant vs SrWO<sub>4</sub> mol fraction



**Fig. 4** The  $Q \times f$  value as a function of SrWO<sub>4</sub> mol fraction



Fig. 5 The  $\tau_{\rm f}$  value as a function of addition of SrWO<sub>4</sub>

be due to the increasing of the B-site ions order because of substituting for the Nb<sup>5+</sup> ions of  $Sr(Mg_{1/3}Nb_{2/3})O_3$  by the  $W^{6+}$  ions of SrWO<sub>4</sub>. Due to the substitution, the charge difference of the B-site ions in Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> increases from 3+, between  $Mg^{2+}$  and  $Nb^{5+}$ , to 4+, between  $Mg^{2+}$ and W<sup>6+</sup>. Also, the ionic radius difference and the B-site ions increase from 0.04 Å, between Mg<sup>2+</sup>(=0.72 Å) and  $Nb^{5+}(=0.68 \text{ Å})$  to 0.14 Å, between  $Mg^{2+}(=0.72 \text{ Å})$  and  $W^{6+}$  (=0.58 Å). So the  $W^{6+}$  ions substitution effectively promoted the charge and ionic radius differences between the B-site ions simultaneously so that the ordering of the B site increased, it is similar to the findings of Galasso and Pyle [4] and Ki Hyun Yoon [5]. In addition, the oxygen vacancies of Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics are eliminated by the 6+ ion substitution of the 5+ ion, Because the oxygen vacancies absorb the electromagnetism wave energy, when the oxygen vacancies were reduced, the  $Q \times f$  values of Sr  $(Mg_{1/3}Nb_{2/3})O_3$  ceramics can increase as well. This can be explained by the following defect Eq. 1:

$$2M_{f}O_{3} \xrightarrow{M_{2}O_{5}} 2M_{Mf}^{\bullet} + 2e^{-} + 5O_{o} + \frac{1}{2}O_{2}$$

$$\frac{1}{2}O_{2} + V_{O}^{\bullet\bullet} + 2e^{-} \rightarrow O_{O}$$

$$2M_{f}O_{3} + V_{O}^{\bullet\bullet} \xrightarrow{M_{2}O_{5}} 2M_{Mf}^{\bullet} + 6O_{O}$$
(1)

Beyond the 0.07 mol SrWO<sub>3</sub>, the  $Q \times f$  values decreases with the increasing SrWO<sub>3</sub>, which is because of the lower  $Q \times f$  value (2,020 at 8.2 G) of SrWO<sub>4</sub> ceramics. Figure 5 illustrates the  $\tau_{\rm f}$  (temperature coefficient of resonant frequency) of the specimens sintered at 1350 °C for 4 h as a function of SrWO<sub>4</sub> mol fraction. According to Fig. 5, with increasing SrWO<sub>4</sub> mol fraction, the SrWO<sub>4</sub> phase( $\tau_{\rm f}$  approximately -50 ppm/°C) increased, the  $\tau_{\rm f}$  value of the specimen became continuously smaller by the volume mixture rule [6], When the mol fraction of SrWO<sub>4</sub> addition is 0.07 mol, the TCF becomes -0.4 ppm/°C.

#### 4 Conclusions

Microwave dielectric properties of x SrTiO<sub>3</sub>–(1-x)Sr(Mg<sub>1/3</sub>  $Nb_{2/3}O_3$  ceramics system have been investigated. As the x value increased from 0.01 to 0.08, the dielectric constants increase from 26.5 to 35.2, while the  $Q \times f$  values gradually decreased and the temperature coefficient of resonant frequency  $(\tau_f)$  varies in a wide range from -23 to 70.6 ppm/°C. When the SrWO<sub>4</sub> was added, the x SrTiO<sub>3</sub>-(1-x)Sr(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub> ceramics is easier to be sintered at lower sintering temperature(1350 °C). In addition, With increasing SrWO<sub>4</sub> mol fraction, the dielectric constant  $\varepsilon_r$ and the  $Q \times f$  values increased initially. Over 0.07 mol SrWO<sub>4</sub>, the dielectric constant  $\varepsilon_r$  and the  $O \times f$  values decreased steadily with increasing SrWO<sub>4</sub>. The TCF value of the specimen decreased with addition of SrWO<sub>4</sub>. As a new microwave dielectric material, 0.025SrTiO<sub>3</sub>-0.975Sr (Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub>-0.07 SrWO<sub>4</sub> can acquire the following microwave properties: a dielectric constant  $\varepsilon_r \sim 30.3$ , a  $Q \times f$ value ~29,500, and a  $\tau_{\rm f}$  value~-0.4 ppm/°C.

Acknowledgements This study is supported by the Chinese Natural Science Fund (No. 60271021) and Chinese National "863" High-tech Project Fund (No. 2001AA-ZB3201-02).

## References

- 1. M.I. Mendelson, J. Am. Ceram. Soc. 52(8), 443 (1985)
- 2. K. Fukuda, R. Kitoh, J. Am. Ceram. Soc. 77(1), 149 (1994)
- 3. C.S. Park, J.H. Paik, S. Nahm, J. Mater. Sci. Lett. 18, 691 (1999)
- 4. F. Galasso, J. Pyle, Inorg. Chem. 2(3), 482 (1963)
- K.H. Yoon, D.P. Kim, E.S. Kim, J. Am. Ceram. Soc. 77(4), 1062 (1994)
- J.M. Wu, M.C. Chang, P.C. Yao, J. Am. Ceram. Soc. 73(6), 1599 (1990)